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If you are viewing this message, it means that we are having trouble loading external resources on our website. If you have a web filter, please make sure that the domains *. kastatic.org and *. Kasandbox.org are allowed. Back to summary of – General Formula – Nomenclature A – Physical Property of Chemical Properties A – Preparations A A – Catalogue Â© Â© Sum: 1) Natural formula: R-NH2Ã© to an amine I (primary) Â) R- NH-R 'a a a a a amine II (secondary) amine III (tertiary) 2) nomenclature: a – primary amine: the name is formed using a fixed which is the name of the alkane (without the final 'e') corresponding to the carbon found e, followed by the position index of the amine group in this chain and that end with "amine." Examples: Formula a name CH3NH2 methanamine CH3-CH2NH2 CH2NAMINE CH3-CH2-CH2NH2 propanamine PROPAN-2-amine 2-METLYPROPAN-2-amine C6H5-CH2NH2 PHE NYLMETHANAMINE Note 1: You can also add add the suffix amine over the name of the alkyl group that is linked to it: CH3-CH2-NH2 ethylamine Note 2: If there are several amine functional groups: NH2- (CH2) 4-NH2 butane-1,4-diamine a – secondary amines: the name is formed from the name of the primary amine corresponding to the carbon group which has the largest number of carbon atoms that are made by the letter "N", followed by a dash and name of the other carbon group attached to the nitrogen atom. N-Mesthanamine (amine II) – tertiary amines: The name is formed from the name of the primary amine corresponding to the carbon group that has the largest number of carbon atoms that occur case by the letter "N" followed by a dash and the name of. one of the other two carbon groups attached to the nit atom hydrogen, followed by a comma and again by the letter "N", followed by a dash and the name of the second carbon group in the nitrogen atom (in the case of identical groups were written two letters "n" selected by coma coma. A guide and the prefix is used Â «diâ» followed by the name of the group). N-Ethyl, N-methylpropanamine Ã, N, N-dimethylethanamine 3) Structure â € "Isomé rich: Ã, Ã, Ã, Ã, Ã, Ã, 3-1) Structure: Derived from the ammonia: it is a pyramid structure. Ammonia n, n-dimethylmetanamine Ã, Ã, Ã, Ã, Ã, Ã,